Acta Crystallographica Section E

## **Structure Reports Online**

ISSN 1600-5368

# (E)-4-(5-Bromo-2-hydroxybenzylideneamino)-1,5-dimethyl-2-phenyl-1H-pyrazol-3(2H)-one

## Guo-Bing Yan, Ming-Hua Yang\* and Liang-Gui Wang

Department of Chemistry, Lishui College, 323000 Lishui, ZheJiang, People's Republic of China

Correspondence e-mail: zilsxvhx@126.com

#### **Kev indicators**

Single-crystal X-ray study T = 293 KMean  $\sigma(C-C) = 0.003 \text{ Å}$ R factor = 0.036 wR factor = 0.095 Data-to-parameter ratio = 14.9

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

In the crystal structure of the title Schiff base compound, C<sub>18</sub>H<sub>16</sub>BrN<sub>3</sub>O<sub>2</sub>, there are three hydrogen bonds which stabilize the molecular and crystal structures.

Received 31 December 2006 Accepted 14 January 2007

#### Comment

Schiff bases showing solvent-dependent UV-vis spectra (solvatochromicity) can be suitable non-linear optical (NLO) active materials (Alemi & Shaabani, 2000). They are also useful in the asymmetric oxidation of methylphenylsulfide and are enantioselective (Kim & Shin, 1999). We report here the synthesis and crystal structure of the title compound (I) (Fig. 1). The crystal structures of similar Schiff bases and of a Cu complex based on such a Schiff base have been reported previously by us (Yan, Yang et al., 2006; Yan, Zheng et al., 2006; Yang, Yan et al., 2006; Yang, Zhang et al., 2007; Zheng et al., 2006).

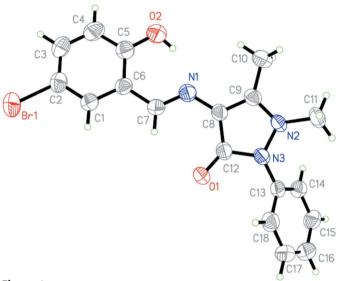
In the crystal structure of (I), there are two intramolecular hydrogen bonds (Table 1) which stabilize the molecular structure, while one intermolecular hydrogen bond stabilizes the crystal structure, forming a chain running along the b axis (Fig. 2).

## **Experimental**

A mixture of 4-amino-1,5-dimethyl-2-phenyl-1,2-dihydropyrazol-3one (2.08 g, 10 mmol), Na<sub>2</sub>SO<sub>4</sub> (3.0 g) and 5-bromosalicylaldehyde (1.32 g, 10 mmol) in absolute ethanol (20 ml) was refluxed under nitrogen for about 12 h, yielding a yellow precipitate. The product was filtered off and washed with ethanol. The crude solid was dissolved in CH<sub>2</sub>Cl<sub>2</sub> (100 ml), washed with water (2 × 10 ml) and brine (10 ml), and dried over Na<sub>2</sub>SO<sub>4</sub>. The remaining solvent was removed under vacuum (yield 92%, 3.1 g). Colourless single crystals of (I) suitable for X-ray analysis were grown by slow evaporation of a  $CH_2Cl_2$ -absolute ethanol (4:1 v/v) solution at room temperature over a period of about a week.

doi:10.1107/S1600536807001742

© 2007 International Union of Crystallography All rights reserved



**Figure 1**The molecular structure of (I), showing the atomic numbering scheme. Non-H atoms are shown as 50% probability displacement ellipsoids.

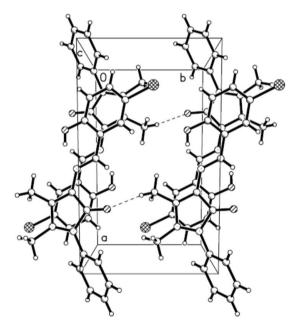


Figure 2 A packing diagram for (I), viewed down the c axis.  $C-H\cdots O$  hydrogen bonds are indicated by dashed lines.

## Crystal data

$C_{18}H_{16}BrN_3O_2$	Z = 4
$M_r = 386.25$	$D_x = 1.520 \text{ Mg m}^{-3}$
Monoclinic, $P2_1/n$	Mo $K\alpha$ radiation
a = 12.1187 (5)  Å	$\mu = 2.45 \text{ mm}^{-1}$
b = 7.0788 (3) Å	T = 293 (2)  K
c = 20.3250 (9)  Å	Block, colourless
$\beta = 104.458 \ (1)^{\circ}$	$0.46 \times 0.37 \times 0.34 \text{ mm}$
$V = 1688.38 (12) \text{ Å}^3$	

#### Data collection

Bruker APEXII area-detector diffractometer  $\varphi$  and  $\omega$  scans Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996)  $T_{\min} = 0.358$ ,  $T_{\max} = 0.435$ 

10580 measured reflections 3273 independent reflections 2248 reflections with  $I > 2\sigma(I)$   $R_{\rm int} = 0.045$   $\theta_{\rm max} = 26.0^{\circ}$ 

#### Refinement

Refinement on  $F^2$   $R[F^2 > 2\sigma(F^2)] = 0.036$   $wR(F^2) = 0.096$  S = 0.943273 reflections 220 parameters

H-atom parameters constrained  $w = 1/[\sigma^2(F_o^2) + (0.0575P)^2]$  where  $P = (F_o^2 + 2F_c^2)/3$   $(\Delta/\sigma)_{\rm max} < 0.001$   $\Delta\rho_{\rm max} = 0.47 \ {\rm e}\ {\rm \mathring{A}}^{-3}$   $\Delta\rho_{\rm min} = -0.47 \ {\rm e}\ {\rm \mathring{A}}^{-3}$ 

**Table 1**Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-\mathrm{H}$	$H \cdot \cdot \cdot A$	$D \cdot \cdot \cdot A$	$D-\mathrm{H}\cdots A$
C10-H10A···O1i	0.96	2.45	3.398 (3)	171
C7−H7···O1	0.93	2.37	3.039 (3)	128
$O2-H2\cdots N1$	0.82	1.87	2.596 (3)	146

Symmetry code: (i) x, y + 1, z.

All H atoms were placed in calculated positions, with C—H = 0.93–0.96 Å and O—H = 0.82 Å, and refined as riding, with  $U_{\rm iso}({\rm H})$  =  $1.2 U_{\rm eq}({\rm C})$  or  $1.5 U_{\rm eq}({\rm methyl}\ {\rm C}$  and O).

Data collection: *SMART* (Bruker, 1998); cell refinement: *SAINT* (Bruker, 1999); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 1998); software used to prepare material for publication: *SHELXTL*.

The authors are grateful to the Natural Science Foundation of Zhejiang Province (grant No. M203052) and the Research Foundation of Lishui University (grant No. FC06002) for financial support.

### References

Alemi, A. A. & Shaabani, B. (2000). Acta Chim. Slov. 47, 363-369.

Bruker (1998). SMART (Version 5.0) and SHELXTL (Version 5.10). Bruker AXS Inc, Madison, Wisconsin, USA.

Bruker (1999). SAINT (Version 6.12). Bruker AXS Inc., Madison, Wisconsis, USA.

Kim, G. J. & Shin, J. W. (1999). Catal. Lett. 63, 83-89.

Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.

Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.

Yan, G.-B., Yang, M.-H. & Zheng, Y.-F. (2006). Acta Cryst. E62, m3481–m3482.
Yan, G.-B., Zheng, Y.-F., Zhang, C.-N. & Yang, M.-H. (2006). Acta Cryst. E62, o5328–o5329.

Yang, M.-H., Yan, G.-B., Zheng, Y.-F. & Zhang, C.-N. (2006). Acta Cryst. E62, o4944–o4945.

Yang, M.-H., Zhang, C.-N., Yan, G.-B., Zheng, Y.-F. & Liu, X.-B. (2007). Acta Cryst. E63, o382–o383.

Zheng, Y.-F., Yan, G.-B. & Gu, Y.-B. (2006). Acta Cryst. E62, o5134-o5135.